DEMO PLANT OF CO₂ SEPARATION AND CONVERSION

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Demo Plant of CO₂ Separation and Conversion

ABSTRACT

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To keep in step with international trend and to reduce the emission of CO₂ at thermal power stations, KEPCO/KEPRI(Korea Electric Power Corporation/Korea Electric Power Research Institute) has implemented environmental protection program to control CO₂ in the flue gas from 1996. The major action to reduce CO₂ emitted from thermal power station is the fabrication of separation and conversion pilot plant.

This paper describes the status of CO₂ separation and conversion technology in KEPCO/KEPRI. In addition, general information on climate change and global warming caused by greenhouse gas will be discussed.



Demo Plant of CO₂ Separation & Conversion

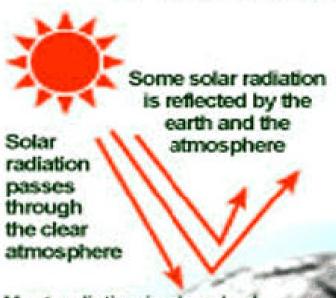
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The Greenhouse Effect



Some of the infrared radiation passes through the atmosphere, and some is absorbed and re-emitted in all directions by greenhouse gas molecules. The effect of this is to warm the earth's surface and the lower atmosphere.

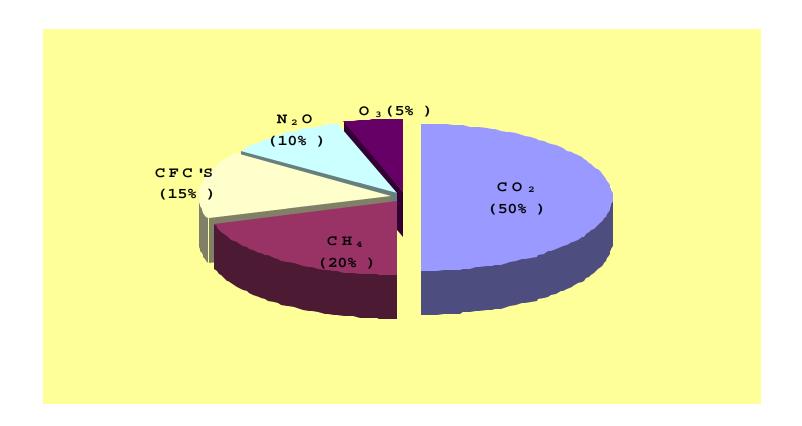
Most radiation is absorbed by the earth's surface and warms it

Infrared radiation is emitted from the earth's surface





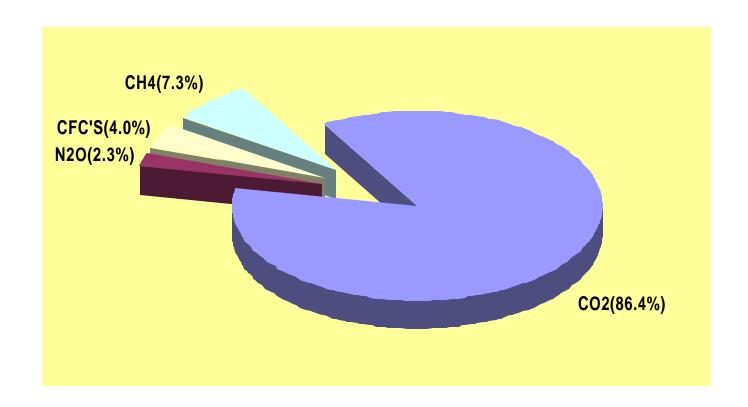
Global Warming Contributed by Greenhouse Gases







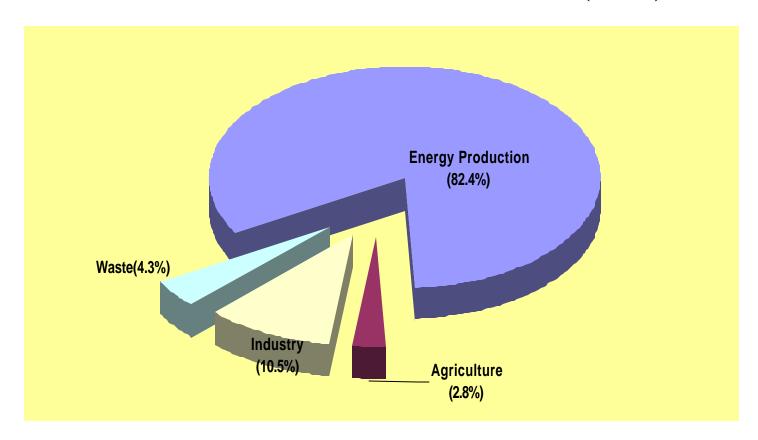
Emission Ratio of GHG in Korea







GHG Emission Source in Korea (1999)







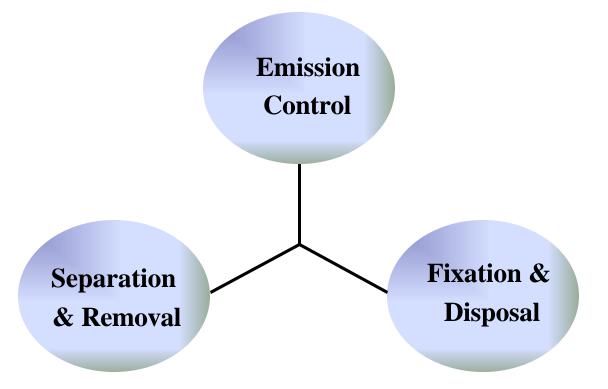
Environmental Changes Caused by Global Warming according to IPCC* report

- 1. Increasement of the surface temperature
- 2. A rise of seawater level
- 3. Agricultural damage
- 4. Ecological adaptation
- 5. Reduction of the forestry area
- 6. Increasement of the disaster
- * IPCC: Intergovernmental Panel on Climate Change





Carbon Dioxide Reduction Technology







CO₂ Reduction Technology at Electric Company

- 1. Elevation of T/P efficiency
- 2. Construction of highly efficient T/P
- 3. Operation of the less CO_2 -emitted T/P (LNG > Oil > Coal)
- 4. Development of new and reusable energy
- 5. Energy saving and the development of power saving facility



Clean Energy Project*

- Title of the Project :
 - A demo-plant study of CO₂ separation and chemical transformation
- Period of the Project :
 - 2000.01 2002. 12 (36 months)
- Main Research Institute: KEPCO (KEPRI)
- Project Team : Environment-Chemistry Group (KEPRI)
- Sub Research Institute: KAIST, KIER, KIST

^{*} Supported by Korean Government



Objectives

- Development of a novel CO₂ separation process for thermal power plants
- Chemical transformation of CO₂ into methanol in pilot plant

Accomplished works

- Development of CO₂ separation technology by absorption
 - Development of a new absorbent :2-amino-2-methyl propanediol (AMPD)
 - Measurement of solubility & absorption rate of absorbents
 - Preparation & operation of lab scale CO₂ separation process (1 Nm₃/hr)

to be continued...





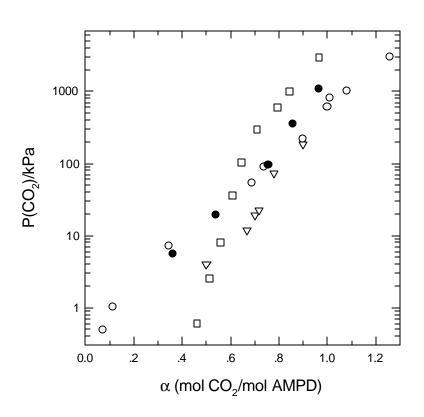
continued

- Improvement of CO₂ absorbents
 - Simulation of separation process using TSWEET
 - Preparation of a wetted wall column absorber to study the absorption ability of absorber types
- Study of the bench scale CO₂ separation process
 - Operation of CO₂ separation process (2 Nm³/hr scale of flue gas)
 - Study on the effect of packing structure
 - Investigation on the thermal degradation of absorbents
 - Study on the scale-up factors for pilot-plant





Development of A New CO₂ Absorbent

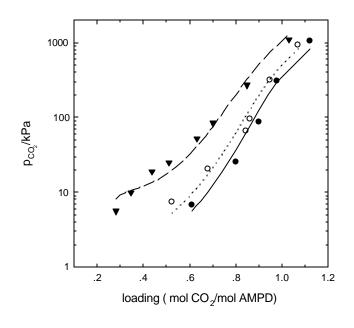


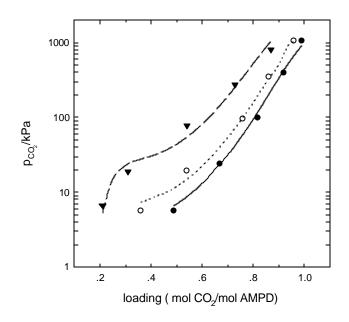
- Solubility of CO₂ in 30 % alkanolamine aqueous solution at 40 :
- () AMPD (this work);
- () MEA (Jou et al., 1995);
- () MDEA (Jou et al., 1994);
- () AMP (Seo and Hong, 1996)





Estimates of CO₂ Solubility by Modified Kent-Eisenberg Model





Solubility in 10 wt % AMPD solution by modelling. $30 \,^{\circ}\text{C} \,(\bullet)$, $40 \,^{\circ}\text{C} \,(\bullet)$, $60 \,^{\circ}\text{C} \,(\blacklozenge)$

Solubility in 30 wt % AMPD solution by modelling. $30 \,^{\circ}\text{C} \,(\bullet)$, $40 \,^{\circ}\text{C} \,(\bullet)$, $60 \,^{\circ}\text{C} \,(\blacklozenge)$







Reaction Mechanism

Primary & Secondary Amine

✓ Absorption : $CO_2 + 2RNH_2 \implies RNHCOO^- + RNH_3^+$

✓ Stripping : $RNHCOO^ \leftrightarrows$ $RNH^- + CO_2$

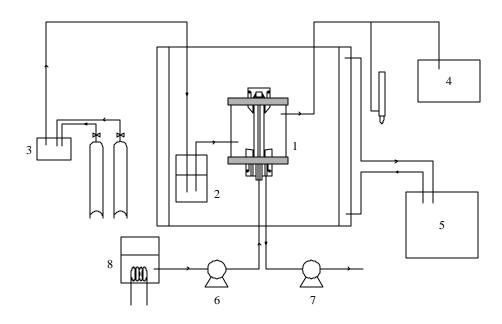
Tertiary Amine

✓ Absorption : $CO_2 + H_2O + R_3N \iff HCO_3^- + R_3N^+H$

✓ Stripping : $HCO_3^ \leftrightarrows$ $OH^- + CO_2$



Wetted Wall Column Absorber



- 1. wetted wall column 2. water saturator
- 3. MFC 4. GC 5. refrigerator/heater
- 6. 7. gear pump 8. absorbent reservoir







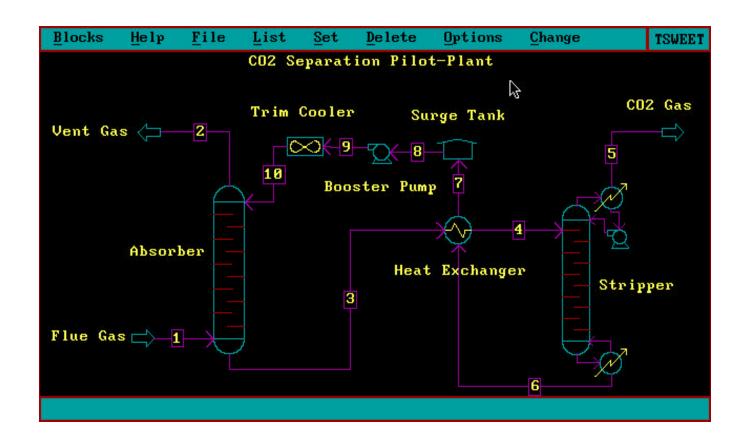
Bench-Scale Apparatus for CO₂ Separation (5Nm³/hr)







Simulation of Pilot-Plant using TSWEET







Condition of the Simulation

• Capacity : 2 CO₂-ton/day

• Recovery ratio of CO₂: 90%

• Absorbent: 30 mass % MEA

• Pressure of flue gas: 150 kPa

Results

- Size of Absorber: Diameter 0.4 m, Height 22 m
- Size of Stripper: Diameter 0.26m, Height 18 m
- Purity of CO₂ to be recovered : 99.95%





CO₂ Separation Pilot Plant at KEPRI

Period of the construction: '01.01 - '01.12 (12 months)

Contractor : Daelim Industrial Co. (Korea)

Site: Seoul LNG thermal power plant #5 (250 MW)

Capacity: 2t-CO₂/day (600 Nm³/hr of flue gas)

Method of capture: Chemical absorption by alkanol amine based solvent

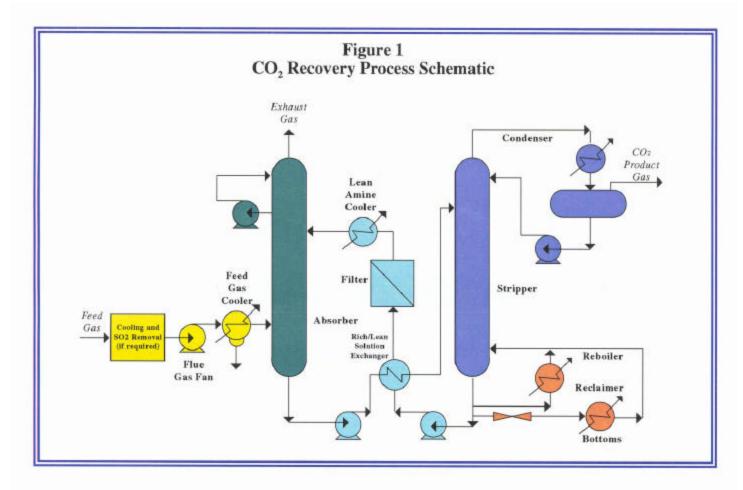
Purity of CO_2 to be recovered : > 99%

Recovery ratio of $CO_2 :> 90\%$





Process Schematic







CAMERE Process

(Carbon dioxide Hydrogenation to Form Methanol via Reverse-Water-Gas-Shift Reaction)

1.
$$CO_2 + H_2 \leftarrow cat$$
 $CO + H_2 + H_2O$

2. Elimination of Water

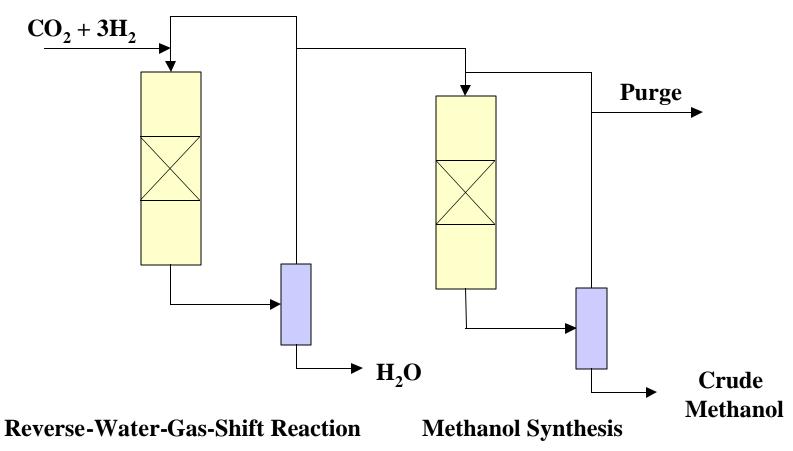
3.
$$CO + H_2 + H_2O \longrightarrow CH_3OH$$

Advantage: Enhancement of process efficiency due to the elimination of H_2O which could be used as an inhibitor





Diagram of CAMERE Process







CAMERE Pilot Plant (5 kg/day)







Future Works

- 1. Development of new absorbents with the high efficiency and stability to surpass classical amine-based absorbents
- 2. Study on the optimal operating conditions of the pilot-plant
- 3. Study on the operating cost reduction
- 4. Study on the thermal degradation, corrosion and foaming of absorbent
- 5. Conversion of CO₂ separated from flue gas into methanol

